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	The objective of this research was the character species which are important because of (a) their or	ccurrence in high-temperature	
COPY	environments, as for example in the vapor over refunction, flames, and propellant burning; (b) the	ractory solids, and in eir relevance to clarification	
8	and/or extension of the basic theory of molecular particles were usually highly reactive or metastable	properties. The molecules	
1 *	usual gas-phase spectroscopic methods. They were	therefore trapped in a solid (continued)	J F
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matrix, usually neon or orgon, at 4°K and investigated by optical and electron-spin-resonance (ESR) spectroscopies. This isolation procedure is known to produce only small perturbations and to yield information pertinent to the gas-phase species. The species studied included boron and bromine atoms, methylene radicals, diatomic boron, beryllium hydroxide, diatomic chlorine anion, carbonyl silene, diazasilene, the first-row transition-metal mono-, di-, and tri-fluorides and their corresponding hydrides and oxides, and a few rare-earth hydrides and fluorides. Vibrational frequencies, electronic transitions, g factors, spin-rotation constants, hyperfine coupling constants, zero-field-splittings, ground-state multiplicities, and perhaps some information about structure, were obtained. The molecules contained from one to seven unpaired electrons.

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8. PUBLICATIONS:

"ESR of Methylene in Neon, Argon, and Krypton Matrices", B.R. Bicknell, W.R.M. Graham and W. Weltner, Jr., J. Chem. Phys. 64, 3319 (1976).

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9. RESEARCH OBJECTIVES

- 1. Characterization of molecular species which are important because of
 - (a) their occurrence in high-temperature environments, as for example in the vapor over refractory solids, and in combustion, flames, and propellant burning.
 - (b) their relevance to clarification and/or extension of the basic theory of molecular properties.

Such species are usually highly reactive and often inaccessible for study by the usual gas phase methods.

2. Exploitation of the matrix-isolation technique to allow the experimental investigation of molecules that are presently impossible or very difficult to study spectroscopically in the gas phase, particularly via electron-spin-resonance (ESR). In this technique the vapor species are trapped in solid neon or argon on a transparent window or dielectric surface at $4^{\rm O}{\rm K}$. The essential virtues of this procedure are (i) the long lifetime of the metastable species in such an inert rigid environment and at such a low temperature, and (ii) the population of only the zero-point vibrational level of the ground electron state of the molecule at $4^{\rm O}{\rm K}$.

10. ACCOMPLISHMENTS

Optical and ESR spectroscopy, and theory, have been applied to characterize a wide variety of small molecules meeting the above criteria. Generally these are the hydride, fluoride, and oxide molecules of the transition and rare-earth metals (usually of high spin), but radicals such as CH₂, BeOH, B₂, and SiCO have also been studied. Abstracts of most of the papers dealing with these molecules are given on the next few pages.

B atoms, B₂ and H₂BO molecules: ESR and optical spectra at 4°K*

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Department of Chemistry, University of Florida, Gainewille, Florida 32611 (Received 3 Match (276)

Elemental boron has been vaportzed and trapped in solid argon at 4-10°K. The Douglas Herzberg transition of B₁ has been observed in absorption at 3300 A at 10°K, indicating that the lower Σ state is the ground electronic state. However, B₁ was not observed via ESR, which is interpreted as support for a ${}^{1}\Sigma_{e}$ ground state with a zero-field splitting greater than about 4 cm⁻¹, rather than ${}^{1}\Sigma_{u}$ favored by *ab initio* calculations. The ESR spectrum of B atoms in solid argon is also detected. It exhibits axial symmetry and almost complete quenching of the orbital angular momentum of the ${}^{2}P_{1/2}$ free atom, with $g_{11} = 2.0014(5)$, g = 1.9645(30). The simple crystal field model has been applied in the interpretation of the g and hyperfine tensors and comparison made with Al and Ga in argon, as studied by Ammeter and Schlosnagle. The ESR spectrum of H₂BO, appearing as an impurity, was observed for the first time, and its magnetic parameters determined

ESR of Cl₂⁻ in various ion pairs at 4 °K and the theory of V centers

J. V. Martinez de Pinillos* and W. Weltner, Jr.

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ESR spectra have been measured of M °Cl₂ ion pairs trapped in solid argon at 4°K, where M = Mg, Ca, Sr, Ba, Li, Na, and K. The derived g and hyperfine tensors of Cl₂ are interpreted using the simple crystal field model of Schoemaker, and the spin densities and crystal field parameters E, are compared with those for V centers in alkali halide crystals. From matrix data, it is proposed that the isotropic hyperfine coupling parameter A_{σ} for interaction with "Cl in pseudo-"free" Cl₂ be taken as 37 ± 1 G. A discussion is given pointing out an approximate linear correlation between the nearest interaction distance (d) and Δg , $(r - A_{\sigma})$ among all the V_{Λ} centers. Constraints imposed by lattices of varying M °/Cl radius ratios cause variations in d and in the s/ρ , character in the antibonding Cl₂ electron, leading to corresponding changes in A_{σ} and inverse changes in Δg . V_{L} centers, with minimum constraints, have maximum values of A_{σ} .

SiC SiN₂, and Si(CO)₂ Molecules: Electron Spin Resonance and Optical Spectra at 4 K

R. R. Lembke, R. F. Fer, ante, and W. Weltner, Jr. *1

Contribution from the Department of Chemistry, University of Florida, Gainesville, Florida 32611. Received August 2, 1976

Abstract: The ${}^{3}\Sigma$ molecules carbonylsilene, SiCO, and diazasilene, SiNN, have been prepared by the vaporization and reaction of silicon atoms with N_{2} or CO and trapped in various matrices at 4 K. Some or all sites in some matrices induced slight bending of the molecules. Isotopic substitution of ${}^{13}C$, ${}^{18}N$, and ${}^{29}Si$ was employed to obtain hyperfine coupling data in the ESR and shifts in the optical spectra. In solid neon, assuming $g=g_{\perp}=g_{s}$, D=2.28 and 2.33 cm $^{-1}$ for SiN $_{2}$ and SiCO, respectively. Hyperfine splittings confirm the CNDO calculated results which indicate that in both molecules the electron spins are largely in the p π orbitals of Si. Optical transitions with vibrational progressions were observed beginning at 3680 and 3108 Å for SiN $_{2}$ and at 4156 Å for SiCO. IR spectra were obtained and stretching force constants calculated. An attempt was made to correlate these vibrational and electronic data with those for CCO and CNN. Annealing an argon matrix containing SiCO to 35 K led to the observation in the IR of ${}^{1}\Sigma$ Si(CO) $_{5}$, a silicon counterpart of carbon suboxide. A corresponding treatment of a SiN $_{2}$ matrix did not produce the N-SiN $_{2}$ molecule, nor was the N₂SiCO molecule observed when both ligands were present.

TiF₂ and TiF₃ Molecules: Electron Spin Resonance Spectra in Rare-Gas Matrices at 4 K

T. C. DeVore and W. Weltner, Jr.*

Contribution from the Department of Chemistry, University of Florida, Gainesville, Florida 32611. Received February 7, 1977

Abstract: TiF₂ and TiF₃ molecules, produced by the vaporization of solid TiF₃ or a mixture of titanium and a solid fluoride, were trapped in neon and argon matrices at 4–40 K. FSR spectra indicated that TiF₃ has a trigonal axis of symmetry with g_{\pm} = 1.8910 (2) and g_{\pm} = 1.9912 (2) and hyperfine tensor components 4 (1i) = -197.9 (2), A_{\pm} (1i) = -178.2 (2), $[A_{\pm}(F)]$ = 47.9 (2), and $[A_{\pm}(F)]$ = 11.5 (2) MHz (in solid neon). It was consided that TiF₃ has a ${}^{2}N_{\pm}$ ground state with the odd electron in a hybridized 4s and 3d $[T_{\pm}^{(1)}]$ orbital perpendicular to the planar (D_{3b}) molecule. The exerted ${}^{2}T_{\pm}$ state has ~2000 cm⁻¹ higher. The ESR spectrum of TiF₃ was that of a nonlinear triplet molecule. Its magnetic parameters are g_{\pm} = 1.9149 (2), g_{\pm} = 1.9229 (2), and g_{\pm} = 1.9880 (2), where the τ axis is parallel to the 1–4 direction. The zero-field-splitting parameters are [D] = 0.0782 (2) cm⁻¹ and [E] = 0.0021 (1) cm⁻¹. The lines were broad, and no hyperfine structure was resolved. Theoretical considerations indicate that the ground state is ${}^{3}B_{\pm}$ with the unpaired spins occupying essentially nonboulding d orbitals on titanium.

YbH and YbD molecules: ESR and optical spectroscopy in argon matrices at 4°Ka)

Richard J. Van Zee, Mikell L. Seely, and W. Weltner, Jr.

Department of Chemistry, University of Florida, Gainesville, Florida 32611 (Received 3 March 1977)

YbH and YbD molecules have been prepared by the reaction of Yb and H(D) atoms during the formation of an argon matrix at 4°K. Yb atom and YbH absorption and emission spectra were observed. The magnetic parameters of YbH were determined from the ESR spectrum of the ${}^{2}\Sigma$ molecules (with Yb nuclear spin I=0 and I=1/2) to be $g_{+}=1.9953(4), g_{+}=1.9402(2), A_{-}(H)=226$ MHz, $A_{-}(H)=224$ MHz, $A_{-}(I^{-1}Yb)(I=1/2)]=5266$ MHz, $A_{-}(I^{-1}Yb)(I=1/2)]=5724$ MHz. The hyperfine parameters indicate that the spin density is less than 20% on the hydrogen and that the bonding is largely Yb'H'. By comparison of experimental parameters with calculated Yb and Yb' data it is deduced that the unpaired electron occupies predominately the $6s\sigma$ orbital on Yb' with smaller contributions of $6p\sigma$ (Yb') and $1s\sigma$ (H').

ESR spectra of the MnO, MnO₂, MnO₃, and MnO₄ molecules at 4°K^{a)}

R. F. Ferrante, J. L. Wilkerson, b W. R. M. Graham, c and W. Weltner, Jr.

Department of Chemistry, University of Florida, Gainesville, Florida 32611 (Received 15 August 1977)

The molecules MnO, MnO₂, MnO₃, and MnO₄ have been prepared by the vaporization and reaction of manganese atoms with O₂, N₂O₃ or O₃ and isolated in various inert-gas matrices at 4°K. ESR has been used to determine magnetic parameters which are interpreted in terms of molecular geometry and electronic structure. MnO is confirmed to have a $\sigma\pi^2\delta^2$, °S? ground state with $g_1=1.990(7)$ (assuming $g_2=g_3$) and a zero-field splitting in accord with the gas phase value D=1.52 cm⁻¹. Hyperfine splittings due to the "Mn (I=5/2) nucleus are A=176(8) and $A_1=440(11)$ MHz. MnO₂ is a linear ⁴S² molecule with probable configuration $\sigma\delta^2$, D=1.13 cm⁻¹ (assuming $g_1=g_1-2.0023$), $A_1=353(11)$, $A_1=731(11)$ MHz. MnO₃ exhibits very large hf splittings $A_1=1772(3)$ and $A_2=1532(3)$ MHz indicative of a M_{22} hybrid M_{12} ground state of D_{36} symmetry. The spectrum of MnO₄ is consistent with a C_{36} molecule distorted from a M_{12} figure state in tetrahedral symmetry by a static Jahn-Teller effect g_1 and A_2 tensors are slightly anisotropic: $g_3=2.0108(8)$, $g_1=2.0007(8)$, $M_{12}=196(3)$ MHz. The electron hole is almost entirely in an oxygen π -bonded orbital with one oxygen atom displaced along its Mn-O bond axis.

Electron Spin Resonance of the Ytterbium Fluoride Molecule at 4 K

R. J. Van Zee, M. L. Seely, T. C. DeVore, and W. Weltner, Jr.

Department of Chemistry, University of Florida, Gainesville, Florida 32611 (Received August 4, 1977)

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The electron spin resonance spectrum of the YbFe'S*) molecule matrix-isolated in solid argon has been observed at 4 K. g tensor and hyperfine tensor components have been measured: $g_{\perp}=1.9954(5), g_{\parallel}=1.9975(5), A_{\parallel}(F)=220(2)$ MHz, $A_{\parallel}(F)=134(2)$ MHz, $A_{\parallel}(F)=1/2$ multiple splittings indicate that the spin density on fluorine is only about 2%, indicating that the molecule is essentially an ion pair, Yb*F. About 80% of the spin is in a Yb* 6s σ orbital and the remainder in 5d σ and 6p σ on the metal ion. From Δg_{\perp} , the spin rotation constant is estimated to be ± 0.0034 cm⁻¹.

High spin molecules: ESR of MnF and MnF2 at 4°K

T. C. DeVore, a) R. J. Van Zee, and W. Weltner, Jr. D)

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ESR spectra of the MnF and MnF, molecules trapped in neon and argon matrices have been observed at 4°K. MnF was found to have a $^{2}\Sigma$ ground state with the following magnetic parameters in solid neon (assuming g=2.002): $g_{\perp}=1.999(1)$, D=-0.0107(1) cm $^{-1}$, $A_{\perp}(A, (Mn))=490(5)$, $A_{\perp}(Mn)=418(1)$, $A_{\perp}(F)=60(1)$, and $A_{\perp}(F)=85(2)$ MHz. MnF is then highly ionic with the spin density on each F⁺ probably less than about 5%. The Mn⁺ ion exhibits about 60% of the scharacter of the free ion. MnF₂ is linear with a $^{+}\Sigma_{g}$ ground state with magnetic parameters in solid neon (assuming g=2.002): $g_{\perp}=1.994(5)$, D=0.370(3) cm $^{-1}$, $A_{\perp}(Mn)=153(6)$, $A_{\perp}(Mn)=124(1)$, $A_{\perp}(F)=19(1)$ MHz. Com₁ arison of these parameters is made with those obtained earlier from crystalline MnF₂ and similar magnetic crystals.

Proc. Symp. on High Temperature

Metal Halide Chemistry (Edited by
D.L. Hildenbrand and D.D.

Cubicciotti) Proceedings Vol. 78-1

pages 187-198 (The Electrochem.

Soc. Inc., 1978).

AN INVESTIGATION OF THE FIRST ROW TRANSITION-METAL FLUORIDE MOLECULES
USING USR SPECTROSCOPY

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ABSTRACT

The ground state electronic configuration and magnetic parameters have been determined for several first row transition metal mono-, di-, and tri-fluoride molecules from their ESR spectra at $^{40}{\rm K}$. The molecules studied have high spin electronic configurations with less than 5% of the free electron spin density tesiding on the fluorine(s), indicative of highly ionic bonding. Theory and experiment have been correlated to establish or predict their ground electronic states and reconstries.

High spin molecules: ESR and optical spectroscopy of MnH ($^{7}\Sigma$) and MnH₂ ($^{6}A_{1}$) at $4^{\circ}K^{a}$)

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MnH and MnH; molecules, and their deuterated counterparts, have been trapped in argon and neon matrices at 4'K and observed via intrared, visible, and electron-spin-resonance spectroscopy. The data for MnH support the gas-phase 2 ground-state assignment and yield the magnetic parameters A(H) < 20, $A_{\infty}(Mn) = 322(0)$, $A_{\infty}(Mn) = 299(2)$ MHz, $g_{\infty} = 2.001(1)$ (assuming $g_{\infty} = g_{\infty}$), D = -0.002(1) cm... The derived MO description is in essential agreement with the *ub initio* calculations of Bagus and Schaeter. Intrared data indicate that MnH₂ is bent at a bond angle of 117 ± 30', and stretching force constants are derived. FSR spectra variations with the matrix used and with isotopic substitution indicate motional effects in sonie matrices. It is concluded that the ground state is ' A_{∞} with D = 0.20(2) cm... and with the probable hyperfine parameters A(H) = 36, $A_{\infty}(Mn) = 73$ MHz, where xy is an average axis perpendicular to the H-H direction. The bent molecule is justified by Walsh-type theory applied to transition-metal dihydrides. There are indications that the MnH₄ molecule may have also been observed.

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THE FOH MOLECULE AT 4 K

A. DENDRAMIS, R. J. VAN ZEE, AND W. WELTNER, JR. Department of Chemistry, University of Florida Received 1978 November 9; accepted 1979 February 5

ABSTRACT

FeH and FeD molecules have been trapped in solid argon at 4 K. From the observed infrared frequencies, corrected for small solid-state effects, the gas-phase vibrational properties are predicted to be $\omega_{e''} = 1764$ (10) and $\omega_{e}x_{e''} = 46$ (5) cm⁻¹. D_0 is found to be 2.0 eV from the Birge-Sponer approximation, but is more likely near 1.7 eV on the basis of the application of that approximation to other diatomic hydrides. Five weak electronic transitions (suggesting low gf values) observed in absorption in the 4000-5000 Å region are probably counterparts of the blue and green band systems in the gas-phase laboratory and stellar spectra. The strongest band occurs at 4190 Å, suggesting that the 4288 Å band studied by Heimer be reinvestigated. Failure to observe an ESR spectrum of FeH in the solid is interpreted as support for an orbitally degenerate ground electronic state, in agreement with the theoretically derived $^6\Delta$ ground state. Subject headings: molecular processes — transition probabilities

CrH and CrH₂ molecules: ESR and optical spectroscopy at 4°K

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CrH and CrD molecules have been trapped in solid argon at 4°K. Infrared spectra, when the large anharmonicity is accounted for, yield predicted gas phase vibrational parameters in the ground state $\omega_r = 1615$ cm⁻¹ and $\omega_r x_e = 17$ cm⁻¹. ESR spectra clearly establish that the molecule has a °S ground state with zero-field-splitting $D_c = 0.34(1)$ cm⁻¹ in solid argon and approximate hyperfine coupling constants 44(H) = 49(5) MHz, $4.(^{51}\text{Cr}) = 53(5)$ MHz, and $g_a : g_c = g_c$. Several "extra lines" (off-principal-axis absorptions) in the ESR support the assignment. CrH₂ (and CrD₂, CrHD) was observed in the IR spectra and may also have been detected in the ESR, which tentatively suggest a S = 2 molecule with $D_c = 0.02$ cm⁻¹.

National Bureau of Standards Special Publication 561, Proceedings of the 10th Materials Research Symposium on Characterization of High Temperature Vapors and Gases held at NBS, Gaithersburg, Maryland, September 18-22, 1978. Issued October 1979.

TRANSITION-METAL MOLECULES AND WALSH'S RULES--RATIONALIZATION OF OPTICAL AND ESR DATA

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The electronic and molecular structures of the transition-metal difluoride, dioxide, and dihydride molecules are rationalized or predicted from proposed molecular orbitals and Walsh-type diagrams based upon optical and ESR spectroscopic data.

ESR of matrix isolated bromine atoms produced in the $H+Br_2$ reaction ^{a)}

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The products of the H(D) + X₂ reaction, where X is Br, Cl, or F, have been trapped in solid argon at 4 °K and observed via ESR. With Br₂ as reactant the observed spectra are attributed to Br atoms electronically quenched in an axial crystal field. The spectra obtained using the other halogens were not clearly attributable to quenched atoms. The ESR of matrix-isolated Br atoms has not been observed previously, [While this paper was being reviewed. H. Muto and L. D. Kispert observed the ESR spectrum of partially quenched Br atoms in x-irradiated N-bromosuccinimed single crystals [J. Chem. Phys. 72, 2300 (1980)]] but their magnetic properties are similar to those recently observed by Iwasaki, Toriyama, and Muto for I atoms quenched in solid xenon. For Br: g = 2.646(1), g = 1.55(1), $\mathcal{U} = 1937(20)$ MHz, $\mathcal{U} = 423(10)$ MHz, and $\mathcal{U} = 100(10)$ MHz. Comparison was made with crude axial crystal field predictions derived using the magnetic parameters of the gas-phase atoms.